

Electrically assisted reverse osmosis for enhancing boron removal in one-pass SWRO desalination

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* Presenting at Session 2.5 Innovations in Desalination, SIWW2022
on 20 April 2022

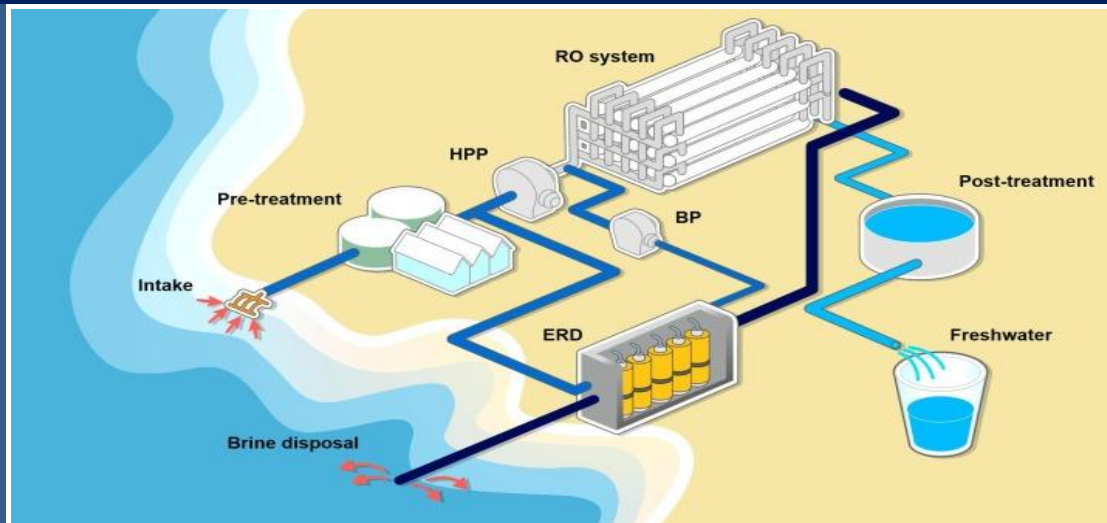


Outline

- Background
- Proposed electrically assisted reverse osmosis (EARO)
- Experiments
- Results

Seawater reverse osmosis desalination

SWRO Desalination



- ✓ Nearly rejects all salt ions
- ✗ Unsatisfied boron rejection

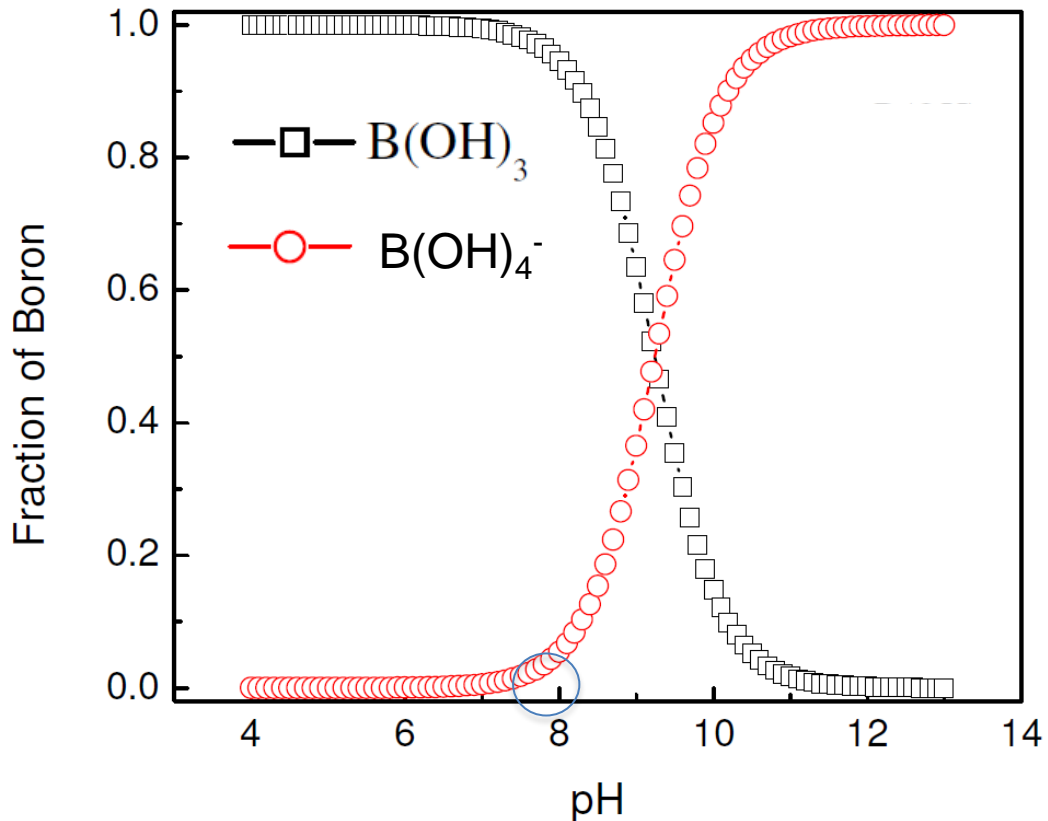


- ✦ WHO stipulated that boron concentration in irrigation water should be lower than **0.5 ppm**

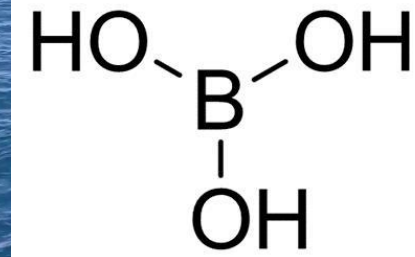
Boron properties

Average 5 ppm in seawater

Boron species as a function of pH



Mainly exists in the form of boric acid in seawater



- Electrically neutral
- Small size (0.244–0.261 nm)

Improve boron removal in SWRO desalination

1) Modification of SWRO membranes

- Plugging
 - Incorporating
- functional additives  Enhance membrane selectivity

Problem: Reduce membrane water permeability

2) Post-treatment of SWRO permeate

- Second RO pass
- Ion exchange resin selective adsorption



Larger size, negatively charged

Problem: Substantial increase in capital cost, footprint, and chemical consumption

3) Pre-treatment of feed seawater

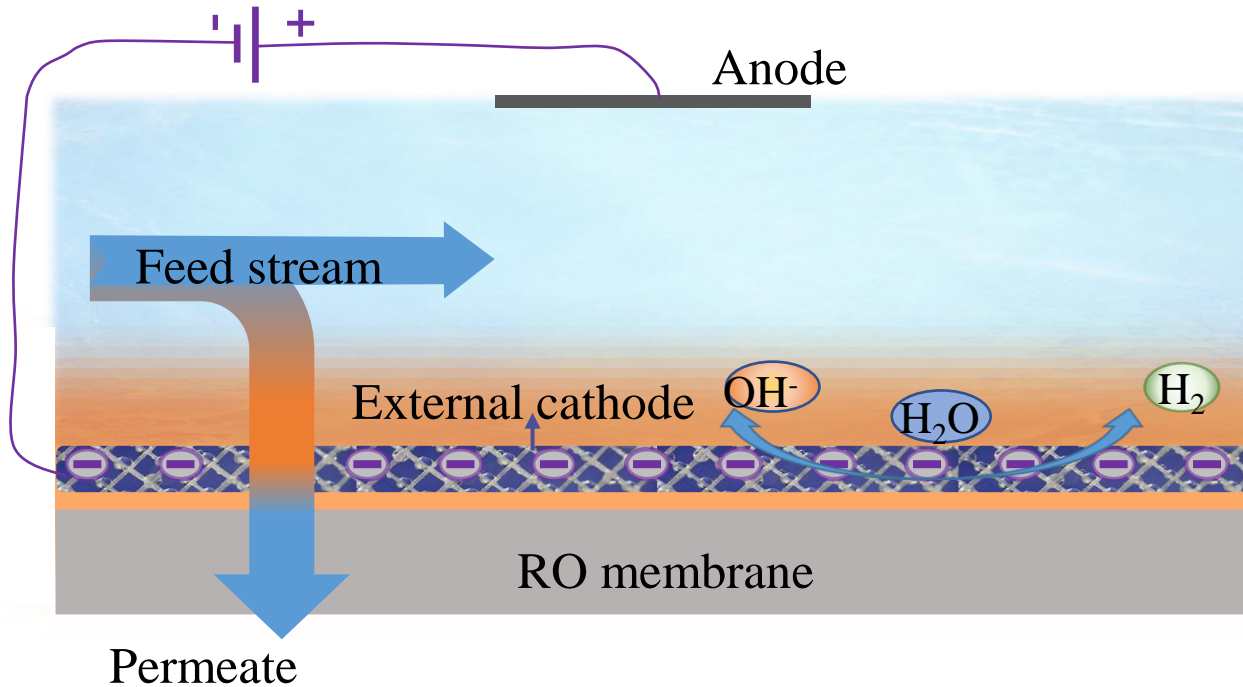
- Elevating feed water pH



Problem: consumes a large amount of alkali for pH adjustment

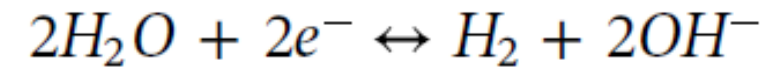
Our method: electrically assisted RO (EARO)

Placing external cathode on membrane surface to elevate local pH instead of bulk feed pH



➔ Maintain the original properties and integrity of the SWRO membrane

Cathodic reaction on membrane surface:



- Boron transport
- Salt transport
- Water flux



Experiments

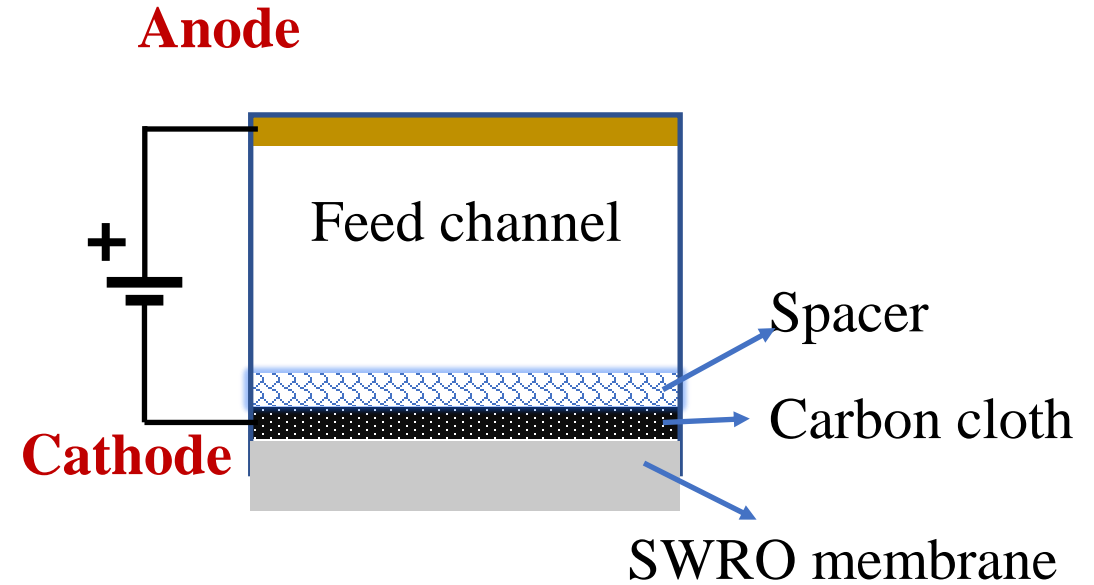
Electrodes → Integrated into feed channel

➤ **Cathode: Porous carbon cloth**

Placed on the top of a commercial SWRO membrane surface in the feed channel

➤ **Anode: Platinum plate**

Placed on the top of feed chamber



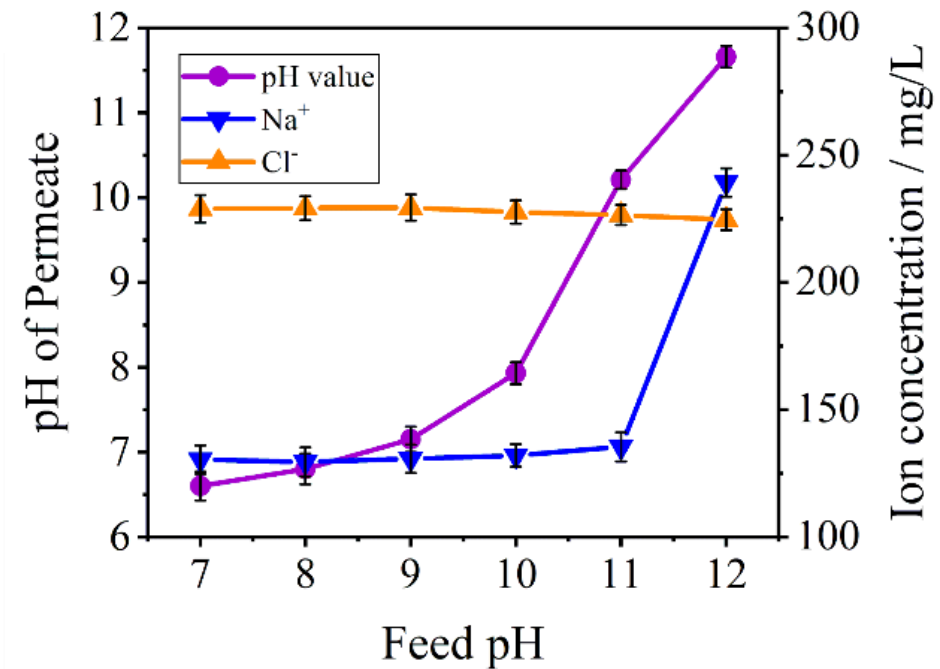
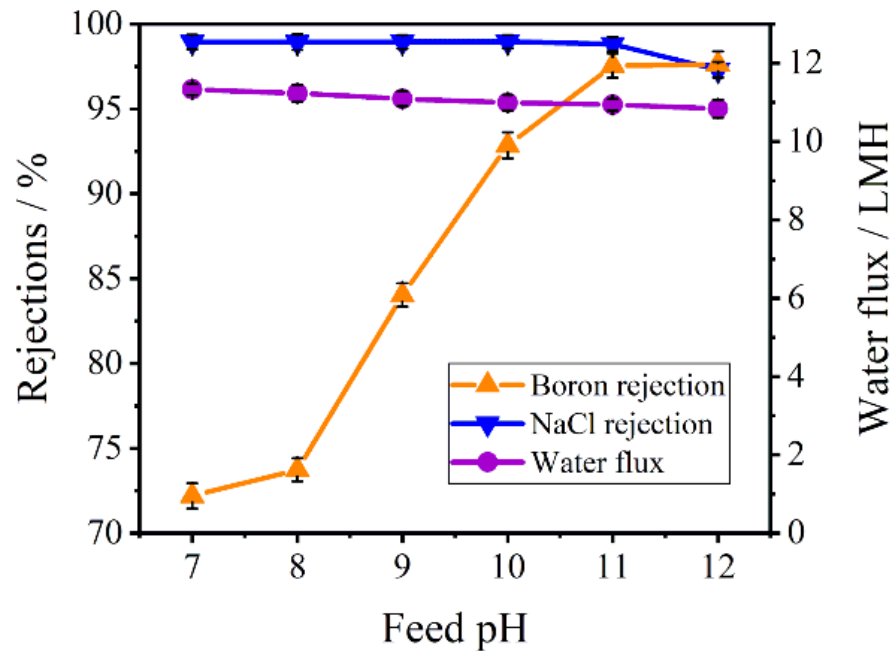
RO pressure: 600 PSI

Feed water

Synthetic salt water with NaCl concentration of 35 g/L (i.e., a typical seawater salt concentration) and boron concentration of 5 mg/L

Performance of conventional RO process

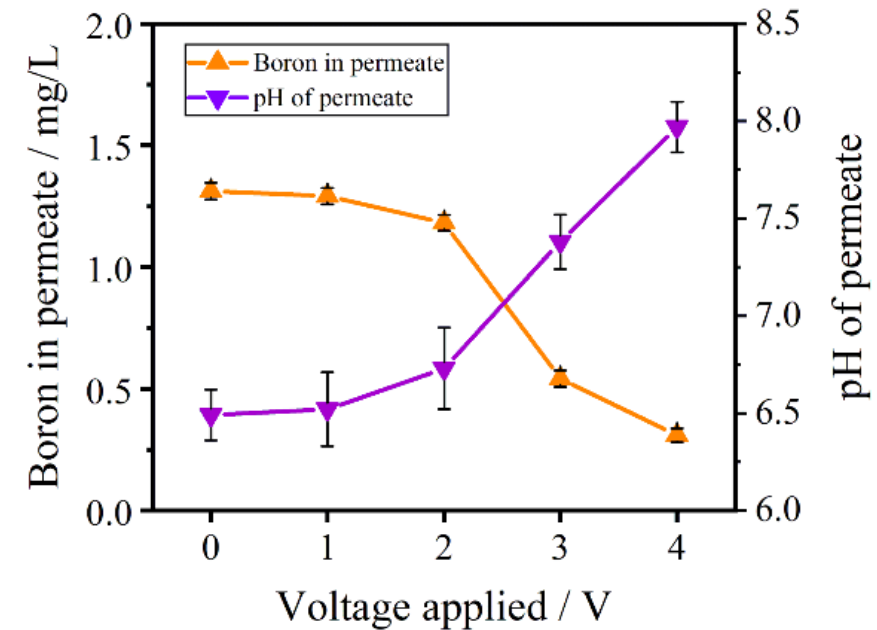
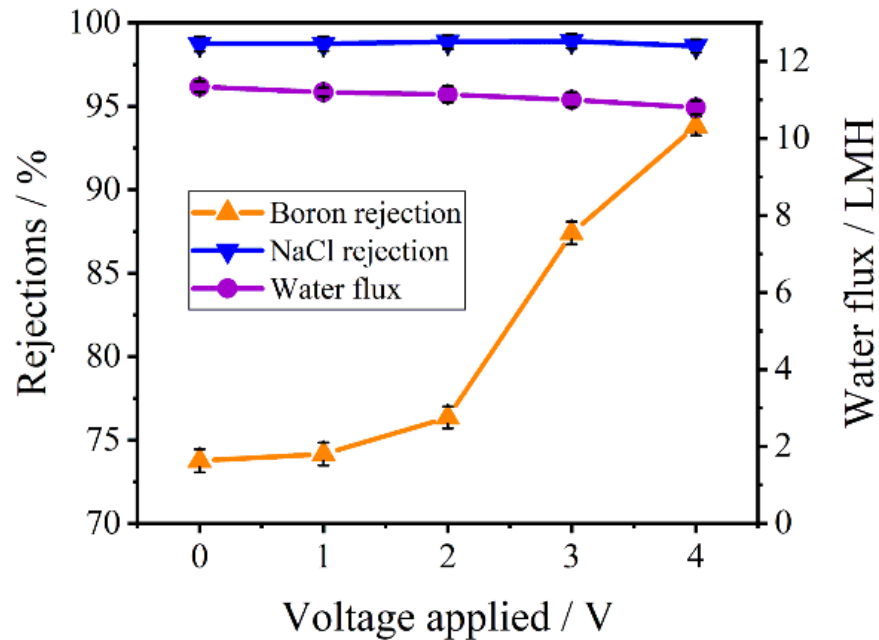
pH value on the performance of conventional RO



- Boron rejection increased with increasing feed pH
- Salt rejection and water flux mildly changed until pH 12

Performance of electrically assisted RO (EARO)

Applied voltage on boron and salt ion transport in EARO

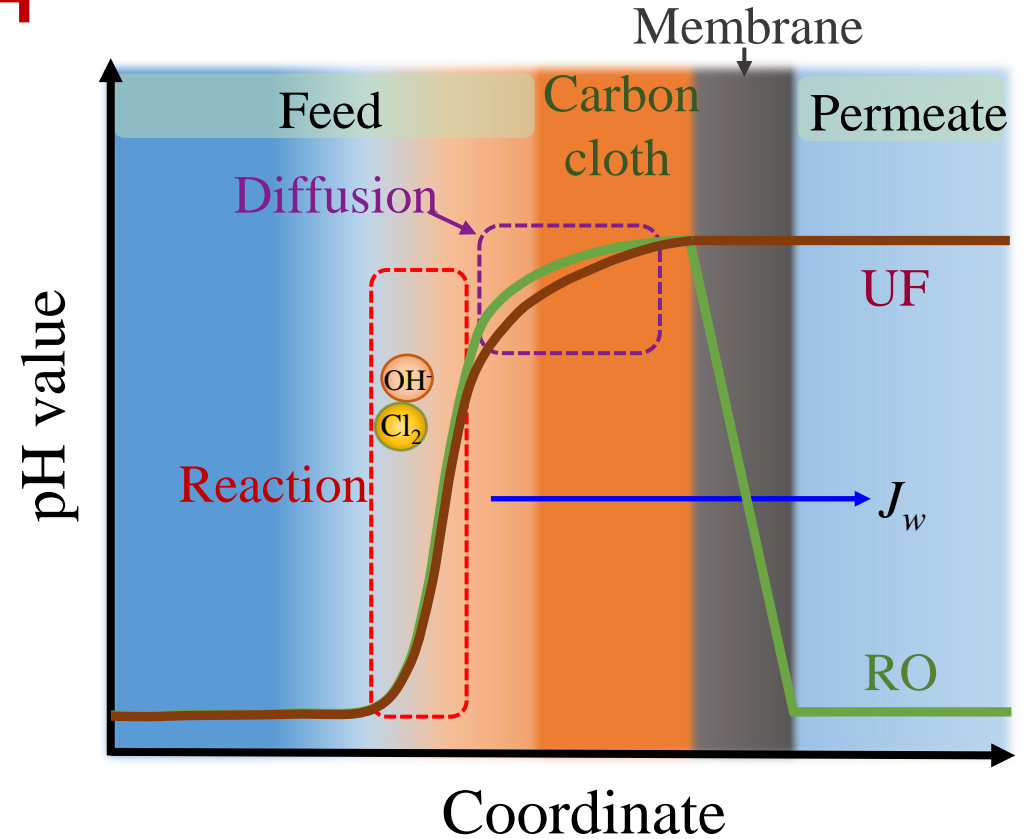
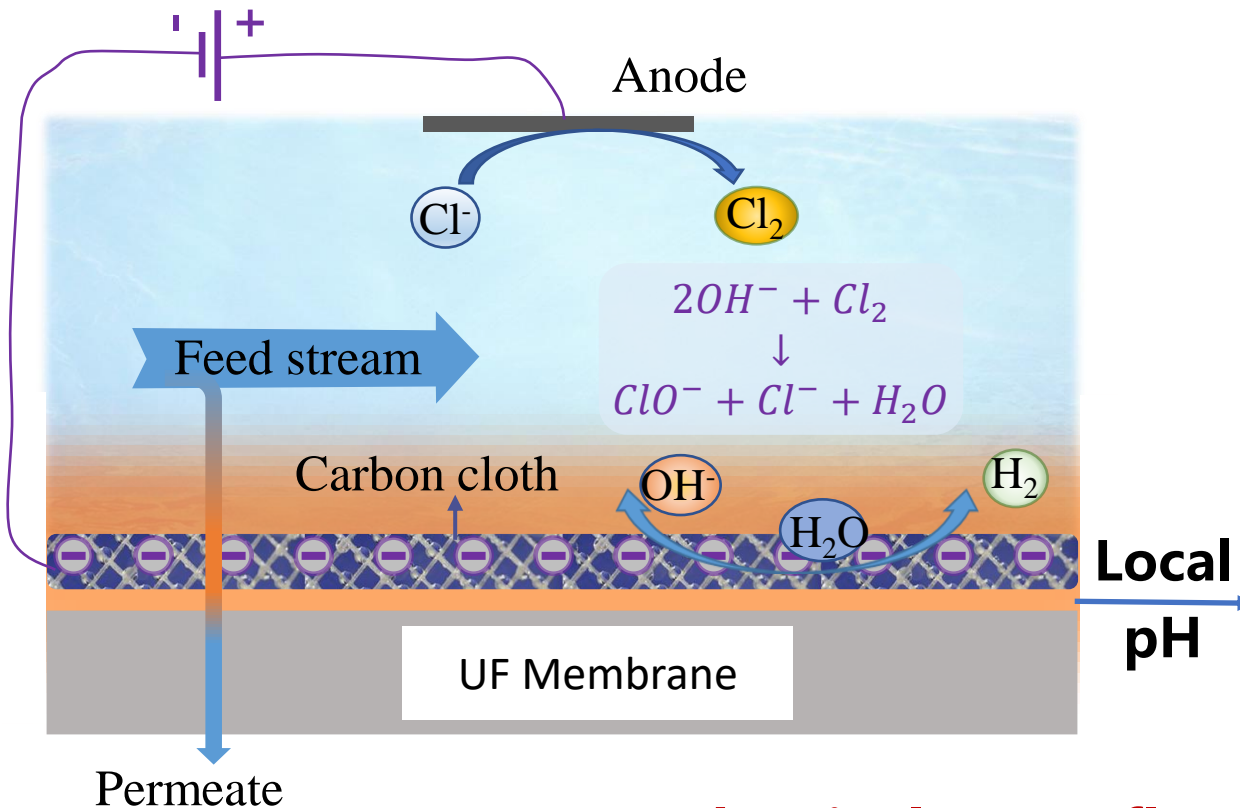


- Boron rejection significantly increased with applied voltage
- Salt rejection and water flux were nearly stable with applied voltage

Determination of local pH in EARO

Boron transport ➤ Enhanced local pH

Local pH can not be measured directly

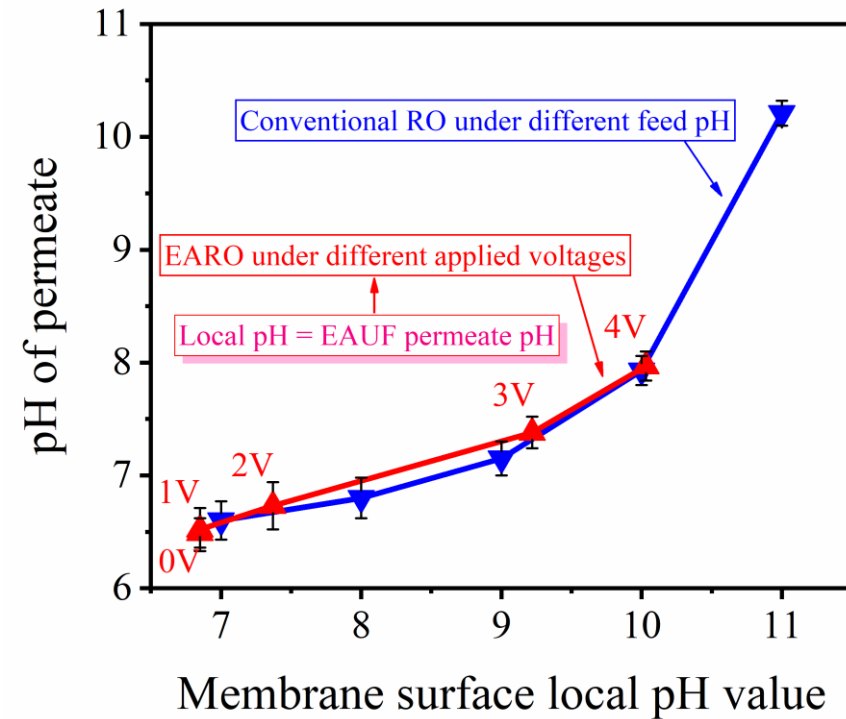
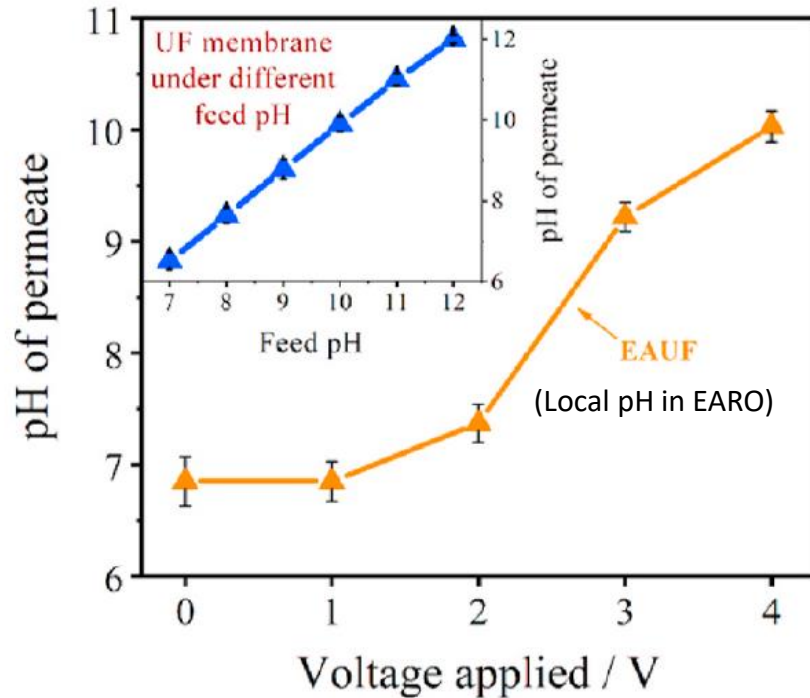


Local pH equals to the permeate pH in EAUF

Identical cross-flow velocity and water flux In EARO and EAUF

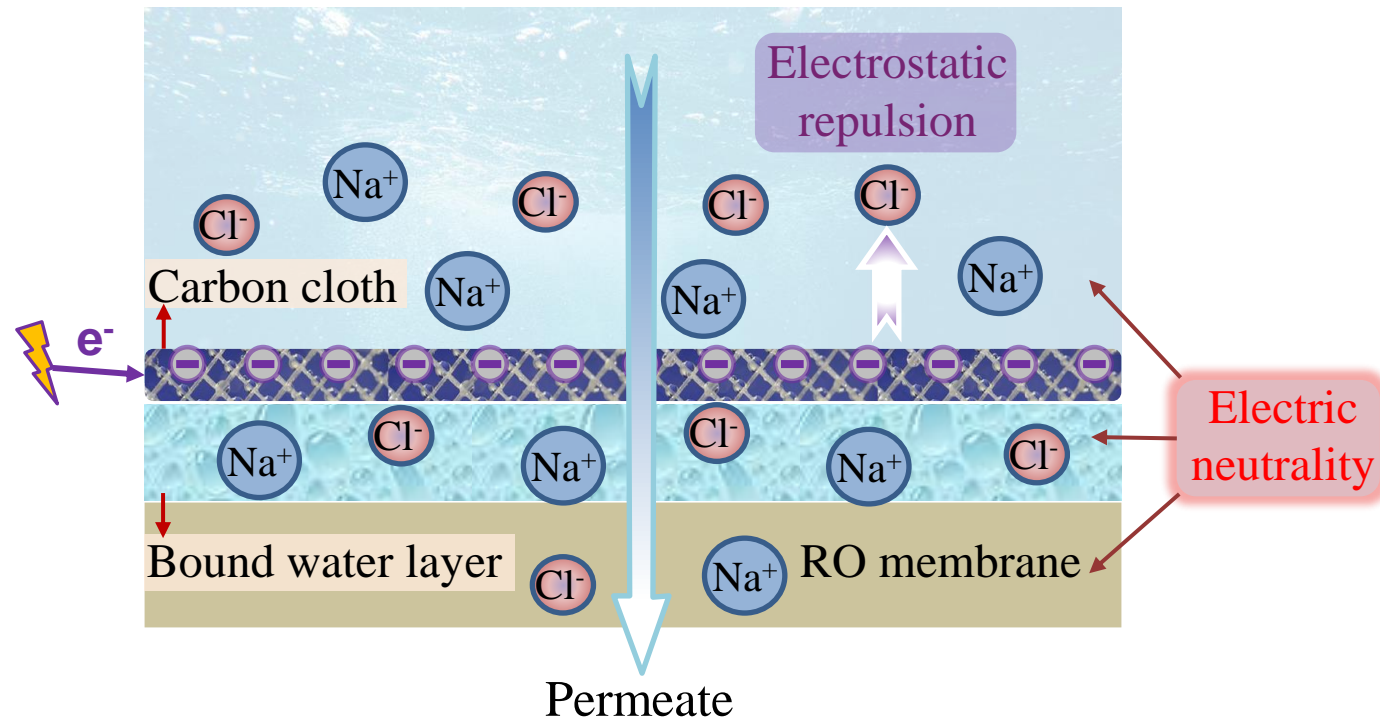
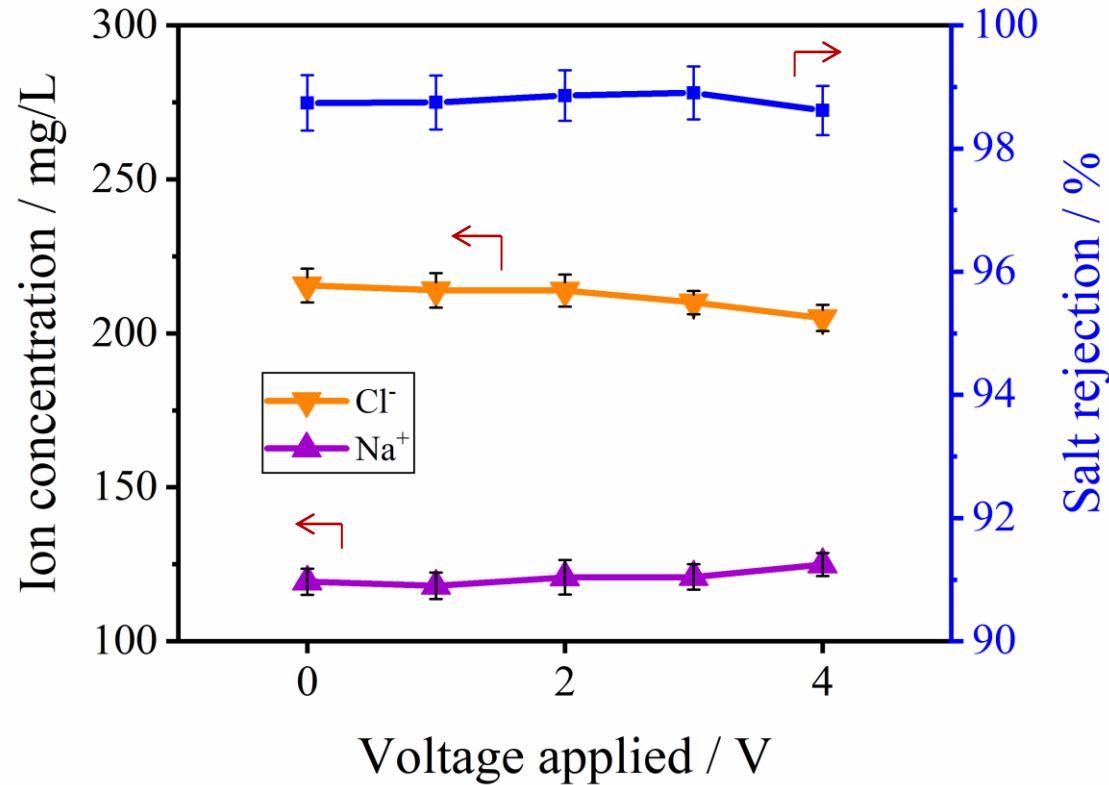
Determination of local pH in EARO

Boron transport ➤ Enhanced local pH



- The local pH in EARO was estimated by the permeate pH in EAUF
- The variations of permeate pH with local pH are similar between EARO and conventional RO

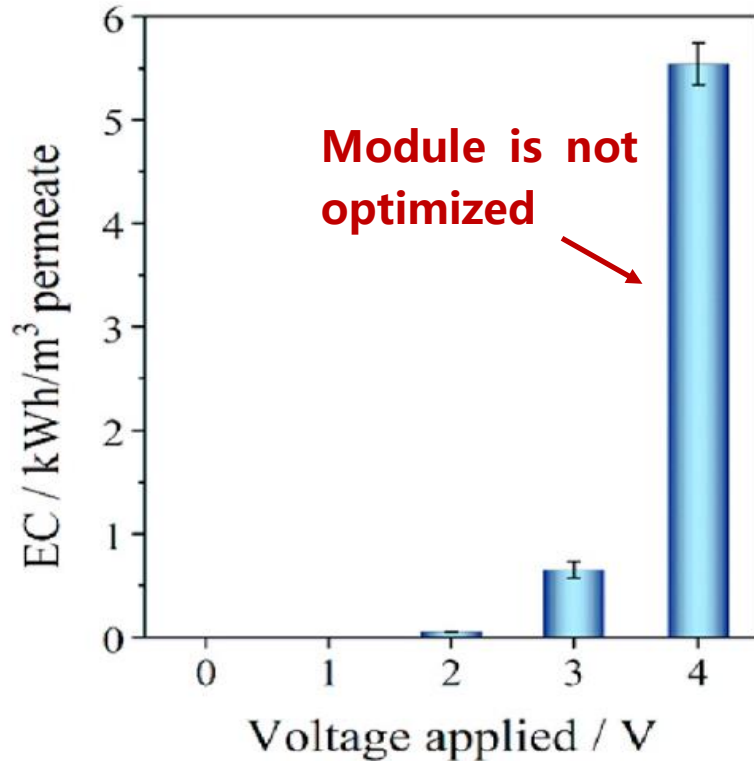
Salt transport in EARO



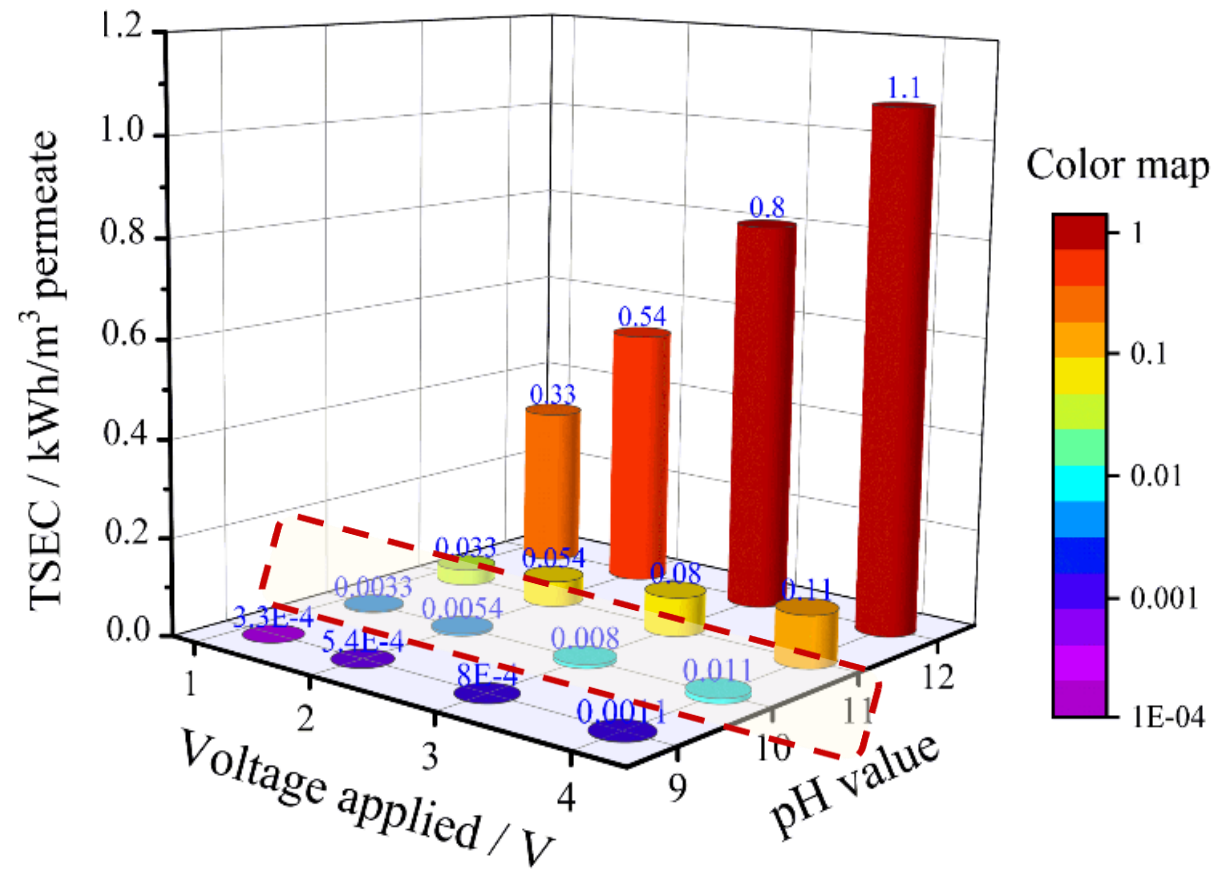
- Salt rejection is almost stable under different voltages
- Slight decrease of permeate Cl⁻ concentration at increasing voltages

Energy consumption of EARO for boron removal

This study



Theoretical specific energy consumption (TSEC)



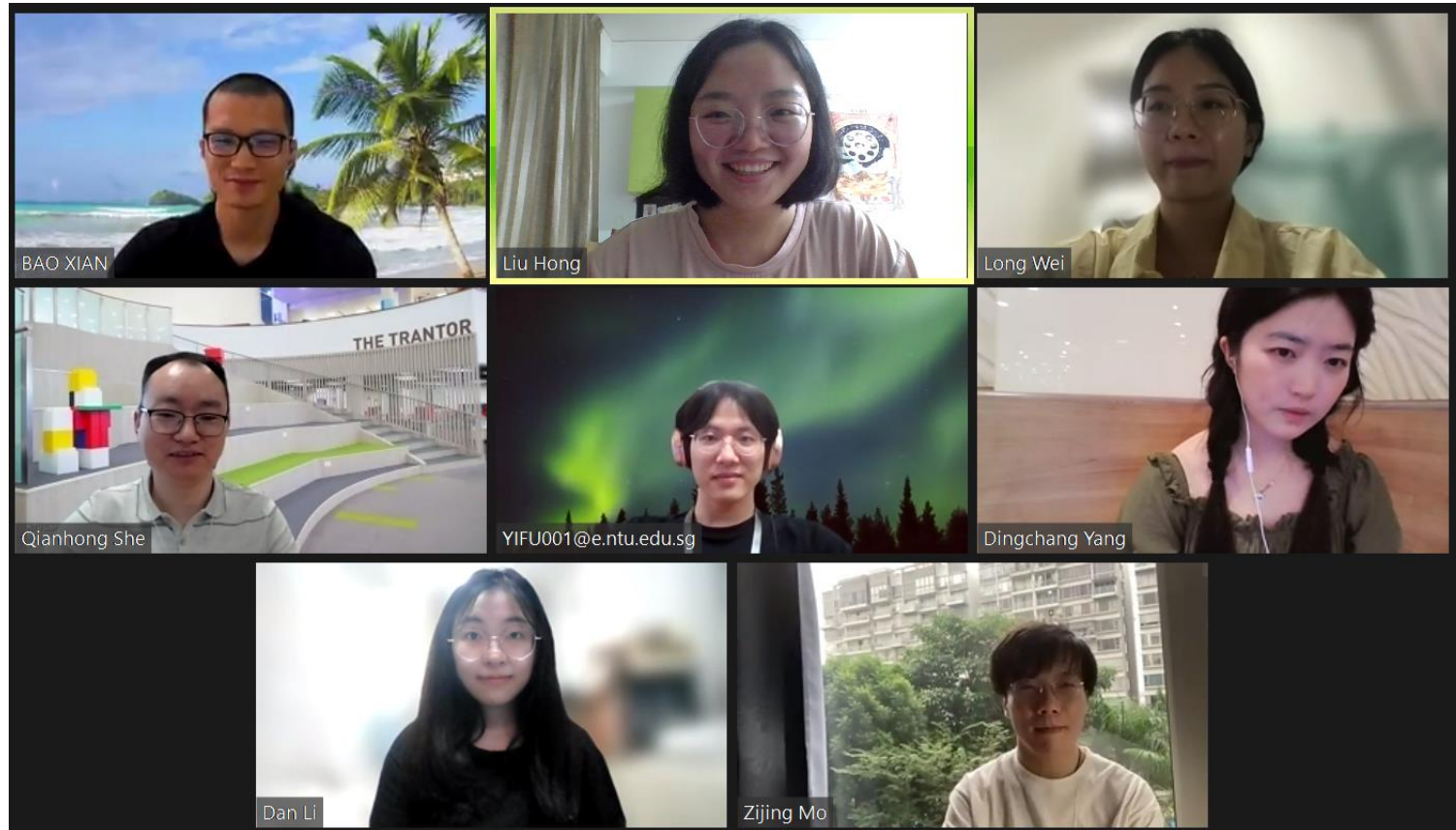
New development: SEC < 0.5 kWh/m³

Conclusions

- A new electrically assisted reverse osmosis (EARO) process was developed by integrating the electrochemical process with reverse osmosis
- Boron rejection increased from ~75% to above 93% when the applied voltage increased from 0 to 4 V in EARO in one-pass SWRO process.
- Salt rejection and water flux was nearly constant under different applied voltages in EARO.
- Low energy consumption can be expected for EARO based on theoretical analysis.
- **EARO is a low energy, no chemical technology for enhancing boron removal in one-pass RO desalination.**
- **Future direction:**
 - to optimize the electrically conductive feed spacer and membrane module design.

Acknowledgements

- Funding agency: Ministry of Education, Singapore, under the Academic Research Fund Tier 1 (RG84/19).
- Group members



Thank you for your attention!

Q&A

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